

## STRUCTURE OF THALFLAVIDINE

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Studying the alkaloids of the roots of *Thalictrum flavum* L., we have isolated yellowish acicular crystals of an optically inactive base  $C_{22}H_{21}NO_6$  with mp 219–220°C (methanol).

In the IR spectrum of the base there are absorption bands at 950 and 1060  $cm^{-1}$  ( $CH_2O_2$ ) and 1730  $cm^{-1}$  (C=O). Its UV spectrum has four absorption maxima, at 254, 263, 296, and 400 nm ( $\log \epsilon$  4.38, 4.49, 3.94, and 3.60), which are characteristic for phenanthrene derivatives [1].

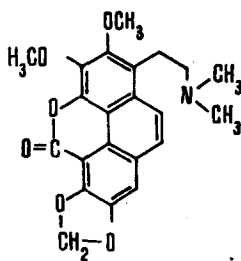
The NMR spectrum of the base shows the signals of protons at ( $\tau$  scale) 7.70 ppm (singlet, 6H) [ $N-(CH_3)_2$ ], two three-proton singlets at 6.18 and 5.96 ppm from two methoxy groups, and a two-proton singlet at 3.71 ppm from a methylenedioxy group.

The base that we isolated proved to be new, and we have called it thalflavidine. On the basis of a study of spectral characteristics and a comparison of them with literature information, thalflavidine can be assigned to the dimethylaminoethylphenanthrene derivatives [1, 2].

The absorption at 1730  $cm^{-1}$  in the IR spectrum of the base shows the presence of an  $\alpha$ -pyrone system in it. The presence of the signal of the protons of the methylenedioxy group in the weak field of the NMR spectrum of thalflavidine (normal values 3.95–4.10 ppm) is due to the action of the carbonyl group and, consequently, the methylenedioxy group is located at  $C_6-C_7$ . Doublets at 2.44 ppm ( $J=9$  Hz), and 2.19 ppm ( $J=9$  Hz) belong to the two ortho-interacting  $C_9$  and  $C_{10}$  protons. A multiplet at 7.50 ppm is due to a methylene group connected with an aromatic ring, and one at 6.74 ppm is due to nitrogen. A singlet at 2.56 ppm is due to the aromatic proton at  $C_8$ .

In the mass spectrum of thalflavidine (taken on an MKh-1303 instrument, 40 V, 0.4 mA, 120°C) there are peaks with  $m/e$  395 ( $M^+$ ), 337 ( $M-58$ )<sup>+</sup>, 322, 279, 58 (100%)  $CH_2=^+N(CH_3)_2$ .

On the basis of these facts, the following most probable structure can be suggested for thalflavidine:



### LITERATURE CITED

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2. N. M. Mollov, Le Nyat Thuan, and P. P. Panov, *Dokl. Bolgarskoi AN*, **24**, No. 8, 1047 (1971).

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